# Structure of Paxisterol, a Sterol from Penicillium paxilli 

Tohru Yasuzawa, Mayumi Yoshida and Hiroshi Sano*<br>Tokyo Research Laboratories, Kyowa Hakko Kogyo Co. Ltd., 3-6-6 Asahimachi, Machida-shi, Tokyo 194, Japan


#### Abstract

The complete structure of paxisterol 1, produced by a Penicillium species, was determined by means of spectral and chemical studies of the parent compound and several of its derivatives. Paxisterol is a unique sterol, having an acetal structure at C -18. Application of 2D-INADEQUATE technique in ${ }^{13} \mathrm{C}$ NMR spectral analysis was useful in this structure determination, where all carboncarbon bonds in the molecule were demonstrated. Absolute stereochemistry was confirmed by the CD spectrum of 3 -keto derivative 10 of compound 1 to be the same as that of cholesterol. An interesting acetal migration reaction was observed during the chemical conversions.


Paxisterol 1, isolated from the culture broth of Penicillium paxilli, has analgesic activity in the mouse acetic acid writhing assay. ${ }^{1}$ We describe here the structure determination of paxisterol, a unique sterol having an acetal structure at C-18.

## Results and Discussion

Compound 1 was obtained as needles, m.p. $198-199^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{25}-60.8^{\circ}\left(c \quad 1.0, \mathrm{CHCl}_{3}\right) ; \lambda_{\max }(\mathrm{MeOH})$ endabsorption. The IR spectrum exhibited absorption bands $v_{\max }\left(\mathrm{CHCl}_{3}\right) 3600$ and $3450(\mathrm{OH}), 1640(\mathrm{C}=\mathrm{C})$ and $1035 \mathrm{~cm}^{-1}$ (C-O-C). The electron-impact mass spectrum, $m / z 442$ ( $M^{+}$, $6 \%$ ), 424, 396, 353, 345, 313 and 93 (100) (Found: $M^{+}, 442.3068$. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{4}$ requires $M, 442.3081$ ), and elementary analysis (Found: C, 76.2; H, 9.2. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{4}$ requires C, $76.33 ; \mathrm{H}, 9.15 \%$ ) indicated that paxisterol 1 had the molecular formula $\mathrm{C}_{28^{-}}$ $\mathrm{H}_{42} \mathrm{O}_{4}$. The proton signal pattern in the ${ }^{1} \mathrm{H}$ NMR spectrum and the similarity of the ${ }^{13} \mathrm{C}$ NMR spectrum of compound 1 with that of $5 x$-cholestan- $3 \beta$-ol indicated that paxisterol 1 possessed a $5 \alpha$-cholestan- $3 \beta$-ol skeleton. The characteristic signals in the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 1 arose from four methyl groups [ $\delta 0.86\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.02\left(6 \mathrm{H}, \mathrm{d}, 26-\mathrm{and} 27-\mathrm{H}_{3}\right)$ and $1.36\left(3 \mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}\right)$ ], three oxygen-bearing methine protons [ $\delta$ 3.93 (br s, $15-\mathrm{H}$ ), $3.54(\mathrm{~m}, 3-\mathrm{H})$ and $3.25(\mathrm{br} \mathrm{d}, 7-\mathrm{H})]$, one acetal proton [ $\delta 5.46(\mathrm{~s}, 18-\mathrm{H})$ ], and one exomethylene group [ $\delta 4.74$ and 4.70 (each br s, $28-\mathrm{H}_{\mathrm{a}}$ and $28-\mathrm{H}_{\mathrm{b}}$ )]. Among these, the methine proton signal at $\delta 3.54$ was assigned to $3-\mathrm{H}$ and the $\beta$ configuration of the 3-hydroxy group was suggested from its coupling pattern. Another methine [ $\delta_{\mathrm{H}} 3.25 ; \delta_{\mathrm{C}} 56.5$ ] appeared to form an epoxide ring together with the quaternary carbon at $\delta_{\mathrm{C}} 60.2$ (from their ${ }^{13} \mathrm{C}$ NMR chemical shifts).

The oxygen bonding to another methine ( $\delta_{\mathrm{H}} 3.93 ; \delta_{\mathrm{C}} 72.8$ ) was found to be involved in the acetal group. Another oxygen of the acetal group was bonded to the quaternary carbon at $\delta_{\mathrm{C}} 85.2$. From this preliminary examination of ${ }^{1} \mathrm{H}$ NMR ${ }^{13} \mathrm{C}$ NMR and ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ COSY spectra it was found that paxisterol 1 contained a $5 \alpha$-cholestan- $3 \beta$-ol skeleton and the three moieties shown in Scheme 1: exomethylene group (A), trisubstituted epoxide group (B) and acetal group (C).

The Location of the Exomethylene Group.-The exomethylene group in paxisterol 1 was characterized by two broad singlets at $\delta 4.74$ and 4.70 in the ${ }^{1} \mathrm{H}$ NMR spectrum and by ${ }^{13} \mathrm{C}$ NMR signals at $\delta 155.6$ and 106.4 (Table 1). Vicinal coupling between terminal methyl protons ( $26-\mathrm{H}_{3}$ and $27-\mathrm{H}_{3}$ ) and $25-\mathrm{H}$, long-range coupling between one of the exomethylene proton $\left(28-\mathrm{H}_{\mathrm{a}}\right)$ and $25-\mathrm{H}$, and an NOE between $28-\mathrm{H}_{\mathrm{a}}$ and $26-\mathrm{H}_{3}$ were



Scheme 1. Partial structures of paxisterol 1.

Table 1. ${ }^{13} \mathrm{C}$ Chemical shifts ( $\delta_{\mathrm{C}}: \mathrm{CDCl}_{3} ; 50.28 \mathrm{MHz}$ ) of the side-chain moieties of paxisterol 1 and 24-methylenecholest-5-en-3 $\beta$-ol.

| Carbon |  | 24-Methylenecholest- <br> 5-en-3 $\beta$-ol |
| :--- | ---: | :--- |
| 20 | 85.2 | 35.6 |
| 21 | 27.3 | 18.7 |
| 22 | 39.6 | 34.5 |
| 23 | 29.5 | 30.9 |
| 24 | 155.6 | 156.6 |
| 25 | 33.9 | 33.7 |
| 26 | 21.9 | 21.8 |
| 27 | 21.9 | 22.0 |
| 28 | 106.4 | 106.0 |

observed. The EIMS showed characteristic fragment ions severed at C-20, C-22, and C-24. These results indicated that the exomethylene group was located at $\mathrm{C}-24$. The ${ }^{13} \mathrm{C}$ chemical shifts of the C-23-through-C-28 atoms of compound 1 were very similar to those of 24 -methylenecholest- 5 -en- $3 \beta$-ol, ${ }^{2}$ indicating that both compounds had the same side-chain at C-20, involving the 24 -exomethylene group (Table 1).

Additional proof was obtained as follows. Hydrogenation of paxisterol 1 with palladium/carbon under hydrogen led to a mixture of $24 R$-and $24 S$-dihydro derivative (2) (Scheme 2) whose ${ }^{13} \mathrm{C}$ NMR signals of the C-20 side-chain resembled those

(3)
(1)
(2)

(4) $\mathrm{R}=\mathrm{Cl}$
(5) $\mathrm{R}=\mathrm{Br}$

(6)



(7) 1 vil

(8)

(10)
(9)

Scheme 2. Reagents and conditions: i, $\mathrm{H}_{2} / \mathrm{Pd}-\mathrm{C} ; \mathrm{ii}, \mathrm{HCl}-\mathrm{THF}, 15 \mathrm{~min}$, room temperature; iii, $\mathrm{PTSA}, \mathrm{CHCl}_{3} ;$ iv, $\mathrm{HCl}-\mathrm{THF}, 1 \mathrm{~h}, 50^{\circ} \mathrm{C}$ for $4 ; \mathrm{HBr}-\mathrm{THF}$, $4 \mathrm{~h}, 50^{\circ} \mathrm{C}$ for 5 ; v, $\mathrm{KOH}-\mathrm{MeOH}$; vi, $\mathrm{LiAlH}_{4}$, THF; vii, $\mathrm{HCl}-\mathrm{THF}$; viii, $\mathrm{Bu}_{3} \mathrm{SnH}, \mathrm{PhMe}$; ix, $\mathrm{Ac} \mathbf{2}_{2} \mathrm{O}$-Py, Jones' reagent; $\mathrm{x}, \mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$.
of 24 -methylcholest- 5 -en- $3 \beta$-ol. ${ }^{3}$ The side-chain structure of compound 1 was thus deduced to be that of a 24 -methylene- $5 \alpha$ -cholestan- $3 \beta$-ol type.

The Location of the Epoxide Group.-The ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum of compound 1 displayed vicinal coupling between the epoxide methine proton ( $\delta 3.25$ ) and a methylene resonating at $\delta$ $c a .1 .6$ and 1.8 (each $1 \mathrm{H}, \mathrm{br}, 6-\mathrm{H}_{2}$ ). Proof that the methylene was attached to the epoxide methine was also established by the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 4 which was thought to be formed by addition of HCl to the epoxide group of compound $\mathbf{1}$, followed by migration of the acetal from $8-\mathrm{OH}$ to $15-\mathrm{OH}$ (Scheme 2). A decoupling experiment on compound 4 showed the connectivity of a chlorine-bearing methine [ $\delta 5.16(1 \mathrm{H}, \mathrm{t}, J$ $2.9 \mathrm{~Hz}, 7-\mathrm{H})$ ] and an adjacent methylene [ $\delta 1.54(1 \mathrm{H}, \mathrm{dt}, J 14.5$ and $2.9 \mathrm{~Hz}, 6-\mathrm{H}_{\mathrm{a}}$ and $2.03(1 \mathrm{H}$, ddd, $J 14.5,12.7$ and $2.9 \mathrm{~Hz}, 6-$ $\mathrm{H}_{\mathrm{b}}$ )]. The acetal-migration reaction was confirmed by the isolation of intermediary product 3 , which was converted into isomer 4 by treatment with toluene- $p$-sulphonic acid (PTSA) in chloroform.

Treatment of 7 -chloride 4 with base failed to produce vinyl compound 6; however, smooth reaction proceeded when bromo-compound 5 was treated with potassium hydroxide in methanol under reflux conditions. The existence of another methine adjacent to the methylene which was bonded to the epoxide methine in paxisterol 1 was deduced from the couplings observed between two vinylic protons at $\delta 5.74(1 \mathrm{H}$, dd, $J 2.9$ and $1.4 \mathrm{~Hz}, 6-\mathrm{H})$ and $6.40(1 \mathrm{H}, \mathrm{dd}, J 1.5$ and 1.4 Hz , 7-H) and an adjacent methine proton in the ${ }^{1}$ H NMR spectrum
of compound 6. Such a situation for a trisubstituted epoxide group is apparently present only in ring $B$ or ring $D$ of the cholestane skeleton.

(11)

Scheme 3. Diagnostic mass fragmentation of keto carboxylic acid 11.
Paxisterol readily underwent oxidation with chromic anhydride in sulphuric acid to give the keto carboxylic acid 11, formed by oxidative cleavage of the epoxide group. ${ }^{4}$ As shown in Scheme 3, in the mass spectrum of compound 11 there are peaks corresponding to the loss of water and a CO group ( $\mathrm{m} / \mathrm{z}$ 454 and 426), indicating the presence of an active hydrogen on the $\gamma$-position to the carboxy group. The fragment ion peak of ring a arising from the cleavage between $\mathrm{C}-9$ and $\mathrm{C}-10$ including


Fig. 1. 2D-INADEQUATE spectrum of paxisterol 1.
the carboxy moiety ( $m / z 168$ ) was also observed. From these findings the carboxymethyl group was determined to be present on $\mathrm{C}-5$ in structure 11, thus confirming the location of the epoxide group to be on $\mathrm{C}-7,-8$ in compound 1 .

The Location of the Acetal Moiety.-Since the signal corresponding to the C-18 methyl group, which is usually found in sterols, was missing in the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of paxisterol 1, the location of the acetal group was estimated to be at C -18. A COLOC experiment of compound 1 showed that the acetal proton [ $\delta 5.46(1 \mathrm{H}, \mathrm{s}, 18-\mathrm{H})$ ] was long-range-coupled both with $\mathrm{C}-20$ at $\delta 85.2$; this was attributed to the long-range coupling with C-21 methyl protons ( $\delta 1.36$ ), and with $\mathrm{C}-15$ at $\delta_{\mathrm{C}}$ 72.8; this was assigned by the observation of long-range coupling between $15-\mathrm{H}$ and $\mathrm{C}-8, \mathrm{C}-10$, and $\mathrm{C}-17$, confirming the connection between $\mathrm{C}-18$ and $\mathrm{C}-20$, and $\mathrm{C}-18$ and $\mathrm{C}-15$, both through an ether linkage (Scheme 4).

(1)

Scheme 4. Long-range couplings around acetal group in paxisterol 1.
The location of the acetal moiety was also strongly supported by the following chemical transformations. Lithium aluminium hydride reduction of the C-7, -8 epoxide group of paxisterol 1 yielded the 8 -hydroxy derivative 7. Acetal migration from 15O to $8-\mathrm{OH}$ in compound 7 readily proceeded under acidic conditions [ HCl tetrahydrofuran (THF)] to form compound 8 (Scheme 2). Compound 8 was also obtained upon treatment of the chloride 4 with tributyltin hydride in hot toluene. As expected, the ${ }^{13} \mathrm{C}$ NMR signal at $\delta_{\mathrm{C}} 71.9$ (C-8) of compound 7 had shifted downfield to $\delta_{\mathrm{c}} 91.8$, and a new proton signal of a secondary hydroxy group [ $\delta 4.25$ (br d, 15-OH)] was observed for compound 8, showing the exchange of the $\mathrm{C}-18$ acetal moiety to the $8-\mathrm{OH}$ group from the $15-\mathrm{O}(\mathrm{H})$, which are close in space. The typical IR absorption for a five-membered-ring
ketone at $1750 \mathrm{~cm}^{-1}$ of compound 9 , which was prepared by selective acetylation of $3-\mathrm{OH}$ in compound 8 , followed by Jones oxidation (Scheme 2), indicated that the newly formed OH group in compound 8, in other words the attached position of the original acetal oxygen in paxisterol 1, was on ring $D$. Furthermore, as the NOE between epoxide methine on C-7 and the acetal oxygen-bonded methine ( $\delta 3.25$ ) was observed in the ${ }^{1} \mathrm{H}$ NMR spectrum of paxisterol 1, the acetal oxygen was deduced to be bonded at $\mathrm{C}-15$. Therefore the location of the acetal moiety was finally decided to be present at C-15/C-18/C20.

Additional skeletal confirmation, including the exomethylene, epoxide and acetal groups, was obtained by 2DINADEQUATE studies of paxisterol 1 as shown in Fig. 1. All bonds between the carbon atoms were observed in this spectrum.

Absolute Stereochemistry.-All- $\beta$-configuration for C-18, $\mathrm{C}-20,15-\mathrm{OH}$ and 7,8 -epoxide was evident from the preceding migration reaction between the acetal present at $\mathrm{C}-15 / \mathrm{C}-18 / \mathrm{C}$ 20 and $8-\mathrm{OH}$ in compound 7. The $\beta$-configuration of the $\mathrm{C}-19$ angular methyl and the $S$-configuration of $\mathrm{C}-20$ were determined by the NOE observed from the $\mathrm{C}-19$ methyl protons ( $\delta 0.86$ ) to $2-\mathrm{H}^{\beta}, 4-\mathrm{H}^{\beta}$ and $11-\mathrm{H}^{\beta}(\delta 1.41, c a .1 .2$, and $c a 1.7$, respectively) and from the C-21 methyl protons ( $\delta 1.36$ ) to 12$H^{\beta}(\delta 2.37)$, respectively, as shown in Scheme 5.

(1)

Scheme 5. Absolute stereochemistry of paxisterol 1.
A CD study was finally carried out to ascertain the absolute stereochemistry of paxisterol 1 . 3-Keto derivative 10 showed a positive Cotton effect in its CD spectrum, similar to that of cholest-5-en-3-one, thus indicating that paxisterol possessed the same absolute stereochemistry as cholesterol.

## Experimental

${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker AM400 spectrometer. Chemical shifts are given from $\mathrm{SiMe}_{4}$ as internal standard, and the coupling constants were reported in hertz. ${ }^{13} \mathrm{C}$ Spectral data are reported in Tables 1 and 2. IR spectra were measured on a Shimadzu IR-27G IR spectrometer. Mass spectra and high-resolution data were recorded on a Hitachi M-80B spectrometer at 70 eV . Optical rotations were measured on a Perkin-Elmer 141 polarimeter, and CD spectra were obtained from a JASCO J-500A spectropolarimeter. M.p.s were taken on a Yanagimoto micro melting point apparatus and are uncorrected. Wakogel C-200 was used for column chromatography, and analytical TLC separations were performed using precoated Silica Gel $60 \mathrm{~F}_{254}$ ( E . Merck) visualized by a spray of $1 \% \mathrm{Ce}\left(\mathrm{SO}_{4}\right)_{2}-10 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ followed by heating.

Catalytic Reduction of Paxisterol 1.-To a solution of paxisterol $1(620 \mathrm{mg})$ in methanol ( 50 ml ) was added $10 \%$ palladium-carbon catalyst ( 450 mg ) and the mixture was stirred

Table 2. ${ }^{13} \mathrm{C}$ NMR data $\left[\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3} ; 50.28 \mathrm{MHz}\right)\right]$ for paxisterol 1 and its derivatives.

| Carbon | 1 | 2 |  | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 38.1 |  |  | 37.6 | 37.5 | 37.4 | 36.2 | 37.8 | 37.8 | 37.5 | 39.4 | 37.2 |
| 2 | 30.7 |  |  | 30.8 | 30.5 | 30.2 | 30.7 | 31.0 | 30.8 | 26.7 | 37.3 | 34.0 |
| 3 | 70.8 |  |  | 71.0 | 70.8 | 70.6 | 71.4 | 71.4 | 71.1 | 73.3 | 210.6 | 210.5 |
| 4 | 37.4 |  |  | 36.8 | 36.9 | 36.8 | 35.1 | 37.8 | 38.0 | 33.7 | 43.9 | 42.6 |
| 5 | 39.8 |  |  | 36.8 | 36.7 | 37.9 | 45.2 | 45.4 | 44.8 | 44.5 | 42.0 | 39.1 |
| 6 | 28.6 |  |  | 36.9 | 33.7 | 34.5 | 126.4 | 24.9 | 25.5 | 25.1 | 28.9 | 30.3 |
| 7 | 56.5 |  |  | 33.0 | 60.6 | 54.7 | 135.0 | 40.3 | 31.8 | 31.7 | 56.1 | 176.7 |
| 8 | 60.2 |  |  | 65.3 | 93.9 | 93.7 | 89.2 | 71.9 | 91.8 | 90.7 | 60.3 | 205.5 |
| 9 | 48.7 |  |  | 75.0 | 48.2 | 48.0 | 55.1 | 53.8 | 54.7 | 54.5 | 48.0 | 55.2 |
| 10 | 33.8 |  |  | 47.3 | 36.5 | 36.6 | 34.5 | 35.8 | 36.3 | 36.3 | 34.0 | 36.5 |
| 11 | 21.7 |  |  | 35.9 | 20.6 | 20.5 | 21.1 | 18.9 | 21.0 | 20.9 | 21.9 | 25.4 |
| 12 | 28.4 | 28.40 | 28.42 | 18.9 | 38.1 | 37.9 | 38.5 | 28.8 | 38.6 | 41.4 | 28.2 | 27.5 |
| 13 | 58.7 |  |  | 28.2 | 63.9 | 63.8 | 65.5 | 57.3 | 64.1 | 60.2 | 58.6 | 63.5 |
| 14 | 54.5 |  |  | 57.6 | 63.5 | 64.4 | 64.4 | 61.6 | 64.4 | 62.8 | 54.3 | 64.3 |
| 15 | 72.8 |  |  | 56.3 | 73.5 | 73.5 | 72.9 | 74.4 | 73.1 | 210.2 | 72.8 | 72.4 |
| 16 | 35.3 | 35.5 | 35.4 | 74.2 | 32.4 | 32.3 | 32.6 | 35.2 | 32.7 | 32.8 | 35.3 | 35.0 |
| 17 | 49.1 | 48.9 | 49.0 | 34.9 | 52.6 | 52.5 | 52.3 | 49.2 | 52.5 | 45.4 | 49.0 | 49.2 |
| 18 | 106.7 | 106.1 |  | 49.2 | 109.2 | 109.3 | 107.7 | 107.1 | 108.3 | 108.1 | 106.6 | 105.1 |
| 19 | 12.8 | 12.8 |  | 106.9 | 12.1 | 12.2 | 11.1 | 12.5 | 12.1 | 12.0 | 12.0 | 16.7 |
| 20 | 85.2 | 85.6 | 85.5 | 12.6 | 88.7 | 88.5 | 88.2 | 85.1 | 88.1 | 87.5 | 85.2 | 86.1 |
| 21 | 27.3 | 27.43 | 27.41 | 85.3 | 24.9 | 24.8 | 24.7 | 27.2 | 24.9 | 24.2 | 27.4 | 27.1 |
| 22 | 39.6 | 39.10 | 39.22 | 27.2 | 37.0 | 36.8 | 36.9 | 39.6 | 36.9 | 36.5 | 39.6 | 39.5 |
| 23 | 29.5 | 28.9 | 29.2 | 39.5 | 28.3 | 28.2 | 28.3 | 29.5 | 28.4 | 28.3 | 29.4 | 30.3 |
| 24 | 155.6 | 39.4 | 39.15 | 29.4 | 155.3 | 155.1 | 155.4 | 155.5 | 155.4 | 155.2 | 155.6 | 155.2 |
| 25 | 33.9 | 31.8 | 31.9 | 155.4 | 33.9 | 33.8 | 33.9 | 34.0 | 33.9 | 34.0 | 33.9 | 33.8 |
| 26 | 21.9 | 18.2 | 17.7 | 34.0 | 21.9 | 21.8 | 21.9 | 21.9 | 21.9 | 21.94 | 21.9 | 21.8 |
| 27 | 21.9 | 20.2 | 20.4 | 21.9 | 21.9 | 21.8 | 21.9 | 21.9 | 21.9 | 21.89 | 21.9 | 21.8 |
| 28 | 106.4 | 15.5 | 15.3 | 21.9 | 106.8 | 106.8 | 106.7 | 106.5 | 106.7 | $106.6$ | 106.4 | 106.5 |
| COMe |  |  |  |  |  |  |  |  |  | 21.4 |  |  |
| COMe |  |  |  |  |  |  |  |  |  | 170.7 |  |  |

for 2 h under hydrogen at room temperature. After the catalyst had been filtered off the solvent was evaporated off under reduced pressure to afford a powder, which was subjected to silica gel column chromatography and eluted with $\mathrm{CHCl}_{3}-$ $\mathrm{MeOH}(10: 1)$ to give a mixture of (24R)- and (24S)-24,28dihydropaxisterol $2\left(440 \mathrm{mg}, 71 \%\right.$ ) as needles, m.p. $234-235^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{22}-55.3^{\circ}\left(c 0.3, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) 3600$ and $3450 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / z 445\left(M \mathrm{H}^{+}, 52 \%\right), 398(100), 300,273$, and 243 (Found: $M^{+}$, 444.3272. $\mathrm{C}_{28} \mathrm{H}_{44} \mathrm{O}_{4}$ requires $M$, 444.3237).

HCl Addition to Paxisterol 1.-To a solution of paxisterol $1(50 \mathrm{mg})$ in THF ( 5 ml ) was added 2m-hydrochloric acid ( 1 ml ), and the mixture was stirred at room temperature for 15 $\min$. The reaction mixture was neutralized with aq. $\mathrm{NaHCO}_{3}$ and evaporated to give a residue, which was dissolved in EtOAc $(50 \mathrm{ml})$ and the solution was washed with water $(50 \mathrm{ml})$. The water layer was extracted twice with EtOAc ( 100 ml ). The combined EtOAc extracts and mother liquor were washed with saturated aq. NaCl and dried with $\mathrm{MgSO}_{4}$. The EtOAc solution was evaporated to afford a residue, which was subjected to silica gel column chromatography and eluted with hexane-EtOAc (3:1) to give 7-chloro-8-hydroxy derivative $3(52 \mathrm{mg}, 96 \%$ ) as needles, m.p. $171-172^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{22}-50.8^{\circ}(c 0.3$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) 3610$ and $3480 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / z 478\left(M^{+}\right.$, $3 \%$ ), 461, 436, 381, 337, 257, 123, 107 and 81 (100) (Found: $M^{+}$, 478.2849. $\mathrm{C}_{28} \mathrm{H}_{43} \mathrm{ClO}_{4}$ requires $M, 478.2848$ ).

HCl Addition followed by Acetal Migration of Paxisterol 1.To a solution of paxisterol $1(1.005 \mathrm{~g})$ in THF ( 30 ml ) was added 2 m -hydrochloric acid ( 1.2 ml ), and the mixture was stirred at $50^{\circ} \mathrm{C}$ for 1 h , neutralized with aq. $\mathrm{NaHCO}_{3}$, and evaporated to give a residue, which was dissolved in $\mathrm{CHCl}_{3}(150$ $\mathrm{ml})$ and washed with water ( 150 ml ). The water layer was extracted twice with $\mathrm{CHCl}_{3}(200 \mathrm{ml})$. The combined $\mathrm{CHCl}_{3}$
extracts and mother liquor were washed with saturated aq. NaCl and dried with $\mathrm{MgSO}_{4}$. The $\mathrm{CHCl}_{3}$ solution was evaporated to afford a residue, which was subjected to silica gel column chromatography and eluted with hexane-EtOAc ( $2: 1$ ) to give 7 -chloro-15-hydroxy derivative 4 ( $745 \mathrm{mg}, 69 \%$ ) as needles, m.p. $169-170{ }^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{22}-43.7^{\circ}$ ( $c 0.2$, $\mathrm{CHCl}_{3}$ ); $v_{\max }\left(\mathrm{CHCl}_{3}\right) 3620$ and $3450 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / 2478\left(M^{+}\right.$, $4 \%$ ), 460 (100), 435, 381, 363, 337, 291, 257, 123, 107 and 81 (Found: $M^{+}, 478.2866 . \mathrm{C}_{28} \mathrm{H}_{43} \mathrm{ClO}_{4}$ requires $M, 478.2848$ ).

Acetal Migration from $15-\mathrm{O}$ to $8-\mathrm{OH}$ in Compound 3.-To a solution of compound $3(630 \mathrm{mg})$ in $\mathrm{CHCl}_{3}$ was added PTSA $(100 \mathrm{mg})$ and the mixture was stirred at room temperature for 4 h , diluted with $\mathrm{CHCl}_{3}$ ( 100 ml ), and neutralized with aq. $\mathrm{NaHCO}_{3}$ checking by universal test paper (Togo Roshi Kaisha, Japan), and washed with saturated aq. NaCl . The $\mathrm{CHCl}_{3}$ solution was dried with $\mathrm{MgSO}_{4}$ and evaporated to give a residue, which was chromatographed on a silica gel column and eluted with hexane-EtOAc ( $2: 1$ ) to give 15 -hydroxy derivative 4 ( $521 \mathrm{mg}, 83 \%$ ).

HBr Addition followed by Acetal Migration of Paxisterol 1.-To a solution of paxisterol $1(750 \mathrm{mg})$ in THF ( 25 ml ) was added 2 M -hydrobromic acid $(1.5 \mathrm{ml})$ and the mixture was stirred at $50^{\circ} \mathrm{C}$ for 4 h . The reaction mixture was neutralized with aq. $\mathrm{NaHCO}_{3}$ checking by universal test paper (Togo Roshi Kaisha, Japan) and evaporated to give a residue, which was dissolved in EtOAc ( 100 ml ) and the solution was washed with water ( 100 ml ). The water layer was extracted twice with EtOAc ( 200 ml ). The combined EtOAc extracts and mother liquor were washed with saturated aq. NaCl and dried with $\mathrm{MgSO}_{4}$. The EtOAc solution was evaporated to afford a residue, which was subjected to silica gel column chromatography and eluted with hexane-EtOAc (2:1) to give 7-bromo-15-hydroxy derivative 5 $\left(620 \mathrm{mg}, 7.0 \%\right.$ ) as needles, m.p. $211-212^{\circ} \mathrm{C}$ (from EtOAc)
(Found: C, 64.0; H, 8.5. $\mathrm{C}_{28} \mathrm{H}_{43} \mathrm{BrO}_{4}$ requires $\mathrm{C}, 64.24 ; \mathrm{H}$, $8.28 \%$ ); $[\alpha]_{\mathrm{D}}^{22}-82.0^{\circ}\left(c 0.3, \mathrm{CHCl}_{3}\right) ; \mathrm{v}_{\max }\left(\mathrm{CHCl}_{3}\right) ; 3620$ and $3450 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / z 523\left(\mathrm{MH}^{+}, 0.2 \%\right), 506,504,481,479,425$ and 81 (100).

Elimination of HBr from Bromide 5.-To a solution of the bromide $5(275 \mathrm{mg})$ in $\mathrm{MeOH}(30 \mathrm{ml})$ was added $\mathrm{KOH}(5 \mathrm{~g})$ and the mixture was heated to reflux for 5 h . The solution was concentrated under reduced pressure to give a residue, which was dissolved in EtOAc ( 50 ml ), and the solution was washed with water ( 50 ml ). The water layer was extracted twice with EtOAc ( 60 ml ). The combined EtOAc extracts and mother liquor were washed with saturated aq. NaCl , dried with $\mathrm{MgSO}_{4}$, and evaporated to afford a residue, which subjected to silica gel column chromatography and eluted with hexane-EtOAc (2:1) to give 6 -ene derivative $6(215 \mathrm{mg}, 93 \%)$ as needles, m.p. $177-$ $178^{\circ} \mathrm{C}$ (from EtOAc) (Found: C, 75.7; H, 9.85. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{4}$ requires $\mathrm{C}, 75.98 ; \mathrm{H}, 9.56 \%$ ); $[\alpha]_{\mathrm{D}}^{22}-56.7^{\circ}\left(c \quad 0.5, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) 3620$ and $3430 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / z 442\left(M^{+}, 3 \%\right), 424$, 406, 328, 312, 300, 283 and 81 (100) (Found: $M^{+}, 442.3107$. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{4}$ requires $M, 442.3081$ ).
$\mathrm{LiAlH}_{4}$ Reduction of Paxisterol 1.-To a stirred solution of paxisterol $1(1.23 \mathrm{~g})$ in THF ( 50 ml ) at room temperature under nitrogen was added $\mathrm{LiAlH}_{4}(250 \mathrm{mg})$ and the mixture was refluxed for 30 min . After the solution had cooled, $\mathrm{MeOH}(5 \mathrm{ml})$ was added followed by water ( 1 ml ). Dil. $\mathrm{HCl}(100 \mathrm{ml})$ was added, and the mixture was extracted twice with EtOAc (200 ml ). The combined extracts were washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and saturated aq. NaCl , dried with $\mathbf{M g S O}_{4}$, and evaporated to yield 8-hydroxy derivative 7 ( 1.21 $\mathrm{g}, 99 \%$ ) as needles, m.p. $206-207^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{22}$ $-25.6^{\circ}\left(c 0.4, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) 3630$ and $3550 \mathrm{~cm}^{-1}(\mathrm{OH})$; $m / z 445\left(M \mathrm{H}^{+}, 60 \%\right), 426(100), 401,329,301,257,123,107$ and 81 (Found: $M^{+}, 444.3249 . \mathrm{C}_{28} \mathrm{H}_{44} \mathrm{O}_{4}$ requires $M, 444.3237$ ).

Acetal Migration from $15-\mathrm{O}$ to $8-\mathrm{OH}$ in Compound 7.-To a solution of compound $7(613 \mathrm{mg})$ in THF ( 50 ml ) was added 2 M hydrochloric acid ( 2 ml ) and the mixture was refluxed for 2 h . After neutralization with aq. $\mathrm{NaHCO}_{3}$ checking by universal pH test paper (Togo Roshi Kaisha, Japan), the solvent was evaporated off under reduced pressure. The residue was dissolved in $\mathrm{CHCl}_{3}(50 \mathrm{ml})$ and the solution was washed with water ( 50 ml ). The water layer was extracted twice with $\mathrm{CHCl}_{3}$ $(100 \mathrm{ml})$. The combined $\mathrm{CHCl}_{3}$ extracts and mother liquor were washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and saturated aq. NaCl and dried with $\mathrm{MgSO}_{4}$. The $\mathrm{CHCl}_{3}$ solution was evaporated to afford a residue, which was subjected to silica gel column chromatography and eluted with hexane-EtOAc (1:1) to give 15-hydroxy derivative $8(415 \mathrm{mg}, 71 \%$ ) as needles, m.p. $171-172^{\circ} \mathrm{C}$ (from EtOAc) (Found: C, 75.4; H, 10.0. $\mathrm{C}_{28} \mathrm{H}_{44} \mathrm{O}_{4}$ requires $\mathrm{C}, 75.63 ; \mathrm{H}, 9.97 \%)$; $[\alpha]_{\mathrm{D}}^{22}-9.3^{\circ}\left(c 0.3, \mathrm{CHCl}_{3}\right)$; $v_{\max }\left(\mathrm{CHCl}_{3}\right) 3600$ and $3440 \mathrm{~cm}^{-1}(\mathrm{OH}) ; m / z 444\left(M^{+}, 8 \%\right), 426$ (100), 408, 401, 329, 301 and 257.

Tributyltin Hydride Reduction of Chloride 4.-To a solution of compound $4(14 \mathrm{mg})$ in toluene $(0.5 \mathrm{ml})$ were added tributyltin hydride ( $20 \mu \mathrm{l}$ ) and $\alpha, \alpha^{\prime}$-azoisobutyronitrile ( 9 mg ) and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 21 h . To the reaction mixture was added $\mathrm{CHCl}_{3}(1 \mathrm{ml})$ and the mixture was stirred for 30 min , then evaporated to afford a residue, which was subjected to silica gel column chromatography and eluted with hexaneEtOAc (2:1) to give compound $8(2.2 \mathrm{mg})$.

Acetylation followed by Jones Oxidation of Compound 8.-A solution of compound $8(50 \mathrm{mg})$ in a mixture of acetic anhydride $(0.5 \mathrm{ml})$ and pyridine ( 1 ml ) under nitrogen was kept for 1 h at room temperature. Water ( 0.5 ml ) was added, and after 30 min
the solvent was evaporated off. To a stirred solution of the residue in acetone ( 2 ml ) was added Jones' reagent $(0.2 \mathrm{ml})$ and the mixture was stirred for 30 min at room temperature, then neutralized with aq. $\mathrm{NaHCO}_{3}$ checking by universal pH test paper (Togo Roshi Kaisha, Japan). Water ( 50 ml ) was added and the mixture was extracted three times with EtOAc ( 150 ml ). The combined extracts were washed with saturated aq. NaCl , dried with $\mathrm{MgSO}_{4}$, and evaporated to afford a yellowish powder, which was subjected to silica gel column chromatography and eluted with hexane-EtOAc (3:1) to give 3-O-acetyl15 -keto derivative $9(45 \mathrm{mg}, 83 \%)$ as needles, m.p. $210-211^{\circ} \mathrm{C}$ (from EtOAc); $[\alpha]_{\mathrm{D}}^{22}+18.9^{\circ}\left(c 0.2, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) 1745$ (ester) and $1725 \mathrm{~cm}^{-1}$ (CO); $m / z 484$ ( $M^{+}, 54 \%$ ), 466, 441, 387, 341 (100), 289, 139, 123 and 107 (Found: $M^{+}, 484.3181$. $\mathrm{C}_{30} \mathrm{H}_{44} \mathrm{O}_{5}$ requires $M, 484.3186$ ).

Pyridinium Chlorochromate (PCC) Oxidation of Paxisterol 1.-To a solution of compound $1(802 \mathrm{mg})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ at room temperature was added pyridinium chlorochromate ( 1.61 g). After 2 h , the reaction mixture was dissolved in EtOAc (100 ml ) and washed with saturated aq. $\mathrm{NaHCO}_{3}$. The water layer was extracted twice with EtOAc ( 200 ml ). The combined EtOAc extracts and mother liquor were washed with saturated aq. NaCl , dried with $\mathrm{MgSO}_{4}$, and evaporated to afford a yellowish residue, which was subjected to silica gel column chromatography and eluted with hexane- $\operatorname{EtOAc}(1: 2)$ to give 3 -keto derivative $10(399 \mathrm{mg}, 50 \%)$ as needles, m.p. $226-227^{\circ} \mathrm{C}$ (from EtOAc) (Found: C, 76.2; $\mathrm{H}, 9.2 . \mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{4}$ requires $\mathrm{C}, 76.33 ; \mathrm{H}$, $9.15 \%$ ); $[x]_{\mathrm{D}}^{22}-29.3^{\circ}\left(c 0.3, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) 1705 \mathrm{~cm}^{-1}$ (CO); $m / z 440\left(M^{+}, 18 \%\right), 422,394,351,343$ (100), 325, 311 and 298 (Found: $M^{+}, 440.2935 . \mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{4}$ requires $M, 440.2925$ ).

Jones Oxidation of Paxisterol 1.-To a solution of paxisterol $1(60 \mathrm{mg})$ in acetone ( 5 ml ) was added Jones' reagent ( 0.2 ml ) and the mixture was stirred for 1 h at room temperature, then evaporated to give a residue, which was dissolved in EtOAc (30 $\mathrm{ml})$, and the solution was washed with water ( 30 ml ). The water layer was extracted twice with EtOAc $(90 \mathrm{ml})$. The combined EtOAc extracts and mother liquor were washed with saturated aq. NaCl , dried with $\mathrm{MgSO}_{4}$, and evaporated to afford a yellowish residue, which was subjected to silica gel column chromatography and eluted with $\mathrm{CHCl}_{3}-\mathrm{MeOH}-$ $\mathrm{AcOH}(30: 1: 0.1)$ to give keto carboxylic acid $11(33 \mathrm{mg}, 52 \%)$ as a powder, $[\alpha]_{\mathrm{D}}^{22}-5.0^{\circ}\left(c 0.2, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) 3200$ and $1710 \mathrm{~cm}^{-1}\left(\mathrm{CO}_{2} \mathrm{H}\right.$ and CO$) ; m / z 472\left(M^{+}, 9 \%\right), 454,426,168$, 123 (85) and 81 (100) (Found: $M^{+}, 472.2835 . \mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{6}$ requires $M, 472.2822$ ).

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